



# Investigation of Heavy Metal Contents in Cow Milk Samples from Area of Merowe-Sudan

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**Abstract:** Cow's milk is considered as one of responsible food sources contamination with heavy metals. The objective of the study were to assess to content of selected heavy metals in fresh cow's milk and its associated human health risks in the food chain in Sudan. Fresh cow's milk samples were collected from two dairy of Merowe city-North Sudan by using Atomic Absorption Spectrophotometer (AAS) was used for the quantitative determination of the concentration of sex heavy metals namely (Cd, Cr, Cu, Pb, Zn and Mn) in samples available, the elements of (Cd 0.007, 0.005, 0.008) (Cr 0.008, 0.009, 0.007) (Cu 0.500, 0.340, 0.460) (Pb 0.009, 0.006, 0.005) (Zn 3.120, 2.000, 1.640) (Mn, 0.120, 0.080, 0.160). These values are compared with Recommended Dietary Allowance (ARD) values and also with corresponding values of different countries available in the literature our result of this paper it is co strong to meager the concentration of seven heavy metals and show an important.

**Keywords:** AAS, Fresh Cow's Milk, Heavy Metals, Contamination, Merowe

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## 1. Introduction

It is well known that increase in industrial and agricultural processes has resulted in increased concentration of metals such as Cu, Zn, Mn, Fe, Cr, Ni, Pb, Cd, Al, Se and Hg in air, water and soil. Such metals are taken in by plants and subsequently get accumulated in their tissues. Thereafter such metals get accumulated in the animals that graze on such contaminated plants and/or in the animals that drink polluted waters and/or inhale polluted air. Metals enter the human body through inhalation of polluted air, ingestion of polluted food and water or absorption through the skin [1]. Thus, large amount of these metals that are taken up by plants and animals subsequently find their way into the food chain. The scientific community is greatly concerned about the ever increasing pollution of the environment, especially about the intake of harmful metals by human beings, plants and animals.

Heavy metals are persistent contaminants is the

environment that can cause serious environmental and health hazards. They are released into the environment from natural and man-made activities. Some heavy metals (like Fe) are essential to maintain proper metabolic activity in living organisms; others (like Pb and Cd) are non-essential and have no biological role [2, 3]. However, at high concentration, they can cause toxicity to living organisms [4].

Lead is recognized as a known neurotoxicant with major public health concern which causes both acute and chronic intoxication [5]. The toxicity may show in the form of anemia, abdominal colic, liver dysfunction, renal damage, peripheral neuropathy in adults, CNS disorders in the form of permanent brain damage in children and in case of extreme lead poisoning, convulsion followed by coma and death, might occur. Moreover, lead has a biological half-life of about 27 years in human bones [6]. Cadmium intoxication in human results in renal damage and dystrophic changes with hypercalcuria, glucosuria, proteinuria and aminoaciduria with hypertension [7], and itai disease which is characterized by

severe pain, soft bones and death may occur as a result of renal failure [8].

## 2. Martial and Methods

### 2.1. Sample Collection and Preparation

Fresh cow's milk from three grazing areas of Merowe metropolis was collected, packed in 50ml bottles and stored in deep freezer at a temperature of -20°C. 20 ml of fresh cow's milk sample was measured into beaker and 20 ml of aqua regia was also added. This was digested by heating the content in the beaker to nearly dryness. After evaporation and cooling, 20ml of distilled water was added to the beaker, stirred and filtered into 50ml volumetric flask and filled to the mark with distilled water. It was then transferred into 50 ml sample bottle labeled and heavy metals were determined using bulk scientific model VER3.94C model Atomic Absorption Spectrophotometer.

### 2.2. Data Assessment

Was collected and ordered all data in tables the concentration of some fresh cow's milk sample in ppm yielding positive results for the occurrence of microelements were transformed into concentration in mg/l.

### 2.3. Chemical and Reagents

All reagents were used of analytical grade deionized water were used to prepare all solution. Nitric acid 65% (Spectrosol, BDH, and England) and hydrogen peroxide 30% (Eurostar Scientific LTD), and concentrated hydrochloric acid. Double distilled water (deionized water) was used for the preparation of all solution. All samples were stored in plastic bottles.

### 2.4. Samples Collection

The polyethylene sampling bottle were soaked in 20% HNO<sub>3</sub> for 24 hours and rinsed with deionised water before collection of raw milk in order to avoid possible contamination. The udder of each cow was washed with distilled water before milking. The three samples of fresh cow's milk of different area were collected from various locations of Merowe and Karrim north Sudan. The study was selected based on the dependence of farmers on live stock production mainly from cows is representative of the whole country as it is considered milk belt in Sudan. A milk sample of 100 ml was collected during morning milking time from each cow and homogenized, and were kept in an ice box. The samples were transported to laboratory and immediately kept a deep freeze (-20°C) until microwave digestion was carried out.

### 2.5. Sample Preparation

A 5ml each aliquot of milk sample was transferred into in 100ml volumetric flask, add 50ml of 24% (w/v) TCA and dilute to volume with deionized water. Shake the samples at 5-min intervals for 30min and filter. Transfer a 5-ml aliquot of

the filtrate to a 50ml volumetric flask, add 1ml of 5% (w/v) lanthanum solution, and make to volume with deionized water.

**Table 1.** Instrumental operating condition for determination of heavy metals in milk sample.

Heavy Metals	Wave length (nm)	SlitWidth (nm)	LampCurrent (mA)	SampleEnergy (eV)
Cd	228.8	0.7	2.0	3.317
Cr	357.9	0.7	2.0	3.567
Cu	324.8	0.7	1.5	3.884
Pb	283.3	0.7	3.0	2.470
Zn	213.9	0.7	2.0	3.148

## 3. Results and Discussion

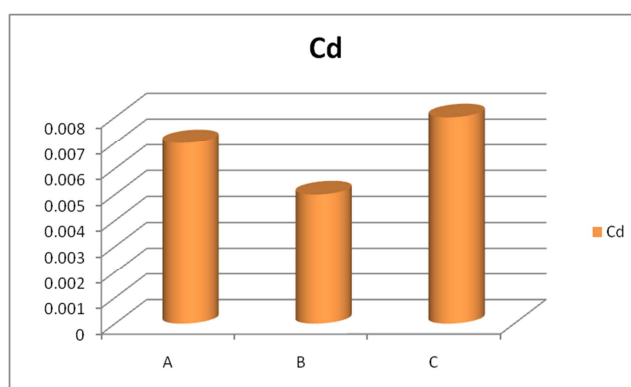
Heavy metals contents varies widely due to many factors such as difference between species, characteristic of manufacturing practices and possible contamination from equipment's during the process among the sex elements analyzed. This could be as a result of low pollution to these metals in the study areas.

**Table 2.** mean concentration of heavy metals in cow milk samples collected from Merowe.

	Cow milk concentration of Heavy metals (mg/kg)					
	Cd	Cr	Mn	Cu	Pb	Zn
A	0.007	0.008	0.120	0.500	0.009	3.120
B	0.005	0.009	0.080	0.340	0.006	2.00
C	0.008	0.007	0.160	0.460	0.005	1.640

### 3.1. Cadmium Concentration

The cadmium concentration in three products samples analyzed was lower than allowed limit (5mg/kg). This could be as a result of low pollution to this metal in the study regular absorption of cadmium causes damage to the proximal renal tubules and calcium phosphorus glucose amino acid and small peptides are lost in the urine once calcium accumulates in tissues it cannot be removed safely by chelation therapy without causing kidney damage. Cadmium affects calcium metabolism and skeletal changes resulting from calcium loss and ends in a decrease in bone mineral density [9, 10].



**Figure 1.** Cadmium concentration.

**3.2. Chromium Concentration**

The concentration of chromium in the milk sample mean (0.008, 0.009, 0.007) this show that the cows milk in the study area is rich in chromium although Cr and other microelements are essential to maintain the metabolic systems of human body they can lead to poisoning at higher level [3].

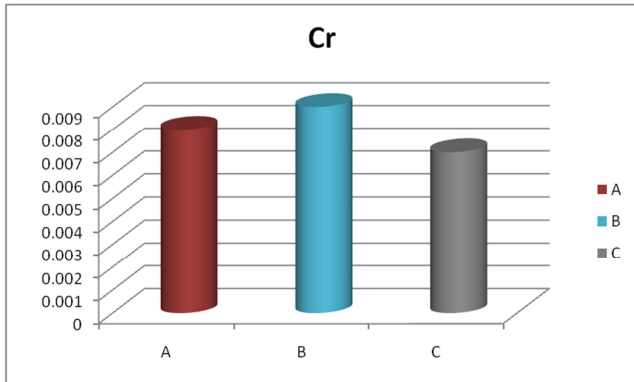


Figure 2. Chromium concentration.

**3.3. Manganese Concentration**

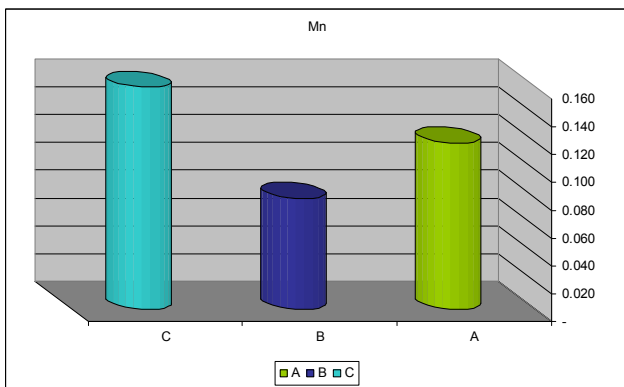


Figure 3. Manganese concentration.

Only a few data on the milk concentration of Mn have been reported. The concentration obtained range between are (0.120, 0.080, 0.160) mg/l in this study is below the accepted limit recommended dietary value of.

**3.4. Copper Concentration**

The possible contamination of milk with copper can occur from animal feed higher copper content in water and also from copper bearing and copper alloys used in equipment contamination of cows milk with copper from brass containers has been cited as the major environmental factor in the etiology of Indian childhood cirrhosis [11, 12]. The concentration of cu in this research work occur between (0.500, 0.340, 0.460) these are below the recommended dietary value stipulated by NRC, 1989. This indicates that cows milk of this study area is poor source of copper. Copper is essential trace element that plays a vital role in the physiology of animals for foetal growth and early post-natal development. Excess copper in the body leads to Wilsons

disease which characterized by deficient of ceruplasmin.

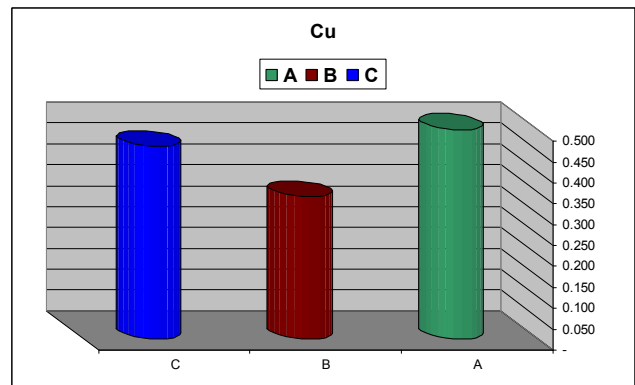


Figure 4. Copper concentration.

**3.5. Lead concentration**

The main sources of lead pollution in the environment include effluents and emissions from industries emissions from vehicles running on leaded petrol, the smoke and dust emissions of coal and gas fired power stations use of lead sheets by roofers as well as the use of paints and anti-rust agents contamination by lead of foodstuffs is caused by the soldered seams of cans and the soldered closures of condensed milk cans the metal caps of wine bottles and also by lead pipes used in drinking water systems. Lead has concentration rang between (0.009, 0.0060, 0.005) mg/L which is above the recommended dietary allowance for adult which is 0.02mg/day. This could be ascribed to the fact that kawo is a cosmopolitan area where a lot human activities were taken part. Lead has no beneficial biological function and is known to accumulate in the body. Lead exposure can cause adverse health effects, especially in young children and pregnant women, since lead is a neurotoxin that permanently interrupts normal brain development.

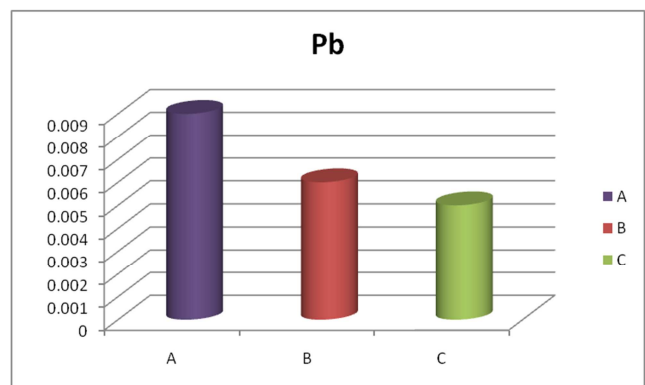


Figure 5. Lead concentration.

**3.6. Zinc Concentration**

In this study was observed in all samples analyzed highest level 3.120 of which was found in the sample from Merowe. This concentration was higher than was observed in Sokoto Metropolis, Nigeria and two-fold lower than [13]. Zinc plays an important role in the immune system but excessive

absorption of zinc suppresses copper and iron absorption which results in decrease in erythrocyte the concentration of Zn is (3.120, 2.00, 1.640).

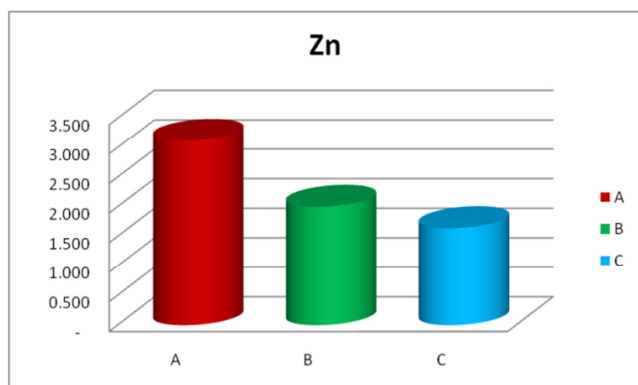


Figure 6. Zinc concentration.

Table 3. Comparison of the elemental concentration (mg/kg) of fresh cow's milk in Sudan.

Country	Cd	Cr	Cu	Pd	Zn	Reference
Japan	1.0	---	100.0	50.0	3000.0	[14]
Germany	0.1	----	49.9	5.5	3730.0	[15]
India	0.1	55.2	56.0	1.6	3100.0	[16, 17]
Spain	----	33.9	9.7	49.3	1419.3	[18]
Poland	15.0	38.0	90.0	20.0	3770.0	[19]
USA	9.7	29.1	19.4	34.0	2235.2	[20]
Sudan	----	----	14.03	----	139	Present study

NR = not reported.

Our results are comparable with values identified by other authors. Table 3 presents a comparison of our values with concentration established in other studies.

## 4. Conclusion

Differences in the concentration of major and minor chemical elements in cow milk depend on many circumstances which are related to geographical location, soil type, nutrition, breed, age, lactation period, geographical location, mineral status of animals. However, sex heavy metals concentration in cow's milk from polluted areas are high compared from non-polluted areas. These differences can be detected even in milk product such as cheese and eggs. Our results confirm attention must be paid to circumstances entailing continuous emission of heavy metals elements into the environment. The concentration of Cd, Cr, Cu, Pb, Mn and Zn were analyzed in cow milk samples collected from three areas of Merowe of Sudan the result revealed that Cd, Cr, Pb, Cu and Zn were detected in all sample and found to be above the stipulated limits of recommended dietary allowance (nrc19989).

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